

# Safety Data Sheet

## 1. IDENTIFICATION




<b>Product Name:</b> Absorptive Glass-Fiber Material Lead Acid Battery <b>Synonyms:</b> AGM Battery	<b>Product Use:</b> Vehicle Electrical System <b>Manufacturer/Supplier:</b> Johnson Controls Battery Group <b>Address:</b> P.O. Box 590 Milwaukee, WI 53201 US
<b>General Information Number:</b> (800)-333-2222 ext. 3138 <b>Contact Person:</b> Industrial Hygiene & Safety Department	<b>Emergency number:</b> CHEMTREC: 800-424-9300

**NOTE:** The Johnson Controls battery is considered an article as defined by 29 CFR 1910.1200 (OSHA Hazard Communication Standard). The information contained in this SDS is supplied at the customer's request for information only.

## 2. HAZARD(S) IDENTIFICATION

Health	Environmental	Physical
Acute Toxicity (Oral, dermal, inhalation) Category 4 Skin corrosion/irritation Category 1A Eye Damage Category 1 Reproductive Category 1A Carcinogenicity (lead) Category 1B Carcinogenicity (acid mist) Category 1A Specific target organ toxicity (repeated exposure) Category 2	Aquatic Chronic 1 Aquatic Acute 1	Explosive Chemical, Division 1.3

### Label Elements:

Health	Environmental	Physical
		
<b>Hazard Statements</b> <b>DANGER!</b> Causes severe skin burns Causes serious eye damage. May damage fertility or the unborn child if ingested or inhaled. May cause cancer if ingested or inhaled. Causes damage to central nervous system, blood and kidneys through prolonged or repeated exposure.	<b>Precautionary Statements</b> Wash thoroughly after handling. Do not eat, drink or smoke when using this product. Wear protective gloves/protective clothing, eye protection/face protection. Avoid breathing dust/fume/gas/mist/vapors/spray. Use only outdoors or in a well-ventilated area. Causes skin irritation, serious eye damage. Contact with internal components may cause irritation or severe burns. Avoid contact with internal acid. Irritating to eyes, respiratory system, and skin.	

May form explosive air/gas mixture during charging. Extremely flammable gas (hydrogen). Explosive, fire, blast or projection hazard.	
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### 3. COMPOSITION / INFORMATION ON INGREDIENTS

INGREDIENTS (Chemical/Common Names):	CAS No.:	% by Wt:
Inorganic Lead/Lead Compounds	7439-92-1	72
Sulfuric Acid absorbed in Glass-Fiber Material	7664-93-9	28

Composition Comments                      All concentrations are in percent by weight.

### 4. FIRST AID MEASURES

**Note: Under normal conditions of battery use, internal components will not present a health hazard. The following information is provided for battery electrolyte (acid) and lead for exposures that may occur during battery production or container breakage or under extreme heat conditions such as fire.**

<b>Inhalation</b>	Sulfuric Acid: Remove to fresh air immediately. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Consult a physician. Lead: Remove from exposure, gargle, wash nose and lips; consult physician.
<b>Skin contact</b>	Sulfuric Acid: Flush with large amounts of water for at least 15 minutes; remove contaminated clothing completely, including shoes. If symptoms persist, seek medical attention. Wash contaminated clothing before reuse. Discard contaminated shoes. Lead: Wash immediately with soap and water.
<b>Eye contact</b>	Sulfuric Acid and Lead: Flush immediately with large amounts of water for at least 15 minutes while lifting lids; Seek immediate medical attention if eyes have been exposed directly to acid.
<b>Ingestion</b>	Sulfuric Acid: Give large quantities of water; Do NOT induce vomiting or aspiration into the lungs may occur and can cause permanent injury or death; consult physician. Lead: Consult physician immediately.

### 5. FIRE FIGHTING MEASURES

<b>Flash Point</b>	Not applicable unless individual components exposed.
<b>Auto ignition Temperature</b>	No data available.
<b>Flammable Limits</b>	LEL = 4.1% (Hydrogen Gas in air) ; UEL = 74.2%
<b>Extinguishing Media</b>	CO <sub>2</sub> ; foam; dry chemical. Do not use carbon dioxide directly on cells. Avoid breathing vapors. Use appropriate media for surrounding fire.
<b>Unsuitable Extinguishing Media</b>	Water
<b>Special Fire Fighting Procedures</b>	Use positive pressure, self-contained breathing apparatus. Beware of acid splatter during water application and wear acid-resistant clothing, gloves, face and eye protection. If batteries are on charge, shut off power to the charging equipment, but note that strings of series connected batteries may still pose risk of electric shock even when charging equipment is shut down.
<b>Unusual Fire and Explosion Hazard</b>	Highly flammable hydrogen gas is generated during charging and operation of batteries. If ignited by burning cigarette, naked flame or spark, may cause battery explosion with dispersion of casing fragments and corrosive liquid electrolyte. Carefully follow manufacturer's instructions for installation and service. Keep away all sources of gas ignition and do not allow metallic articles to simultaneously contact the negative and positive terminals of a battery. Follow manufacturer's instructions for installation and service.

## 6: ACCIDENTAL RELEASE MEASURES

<b>Protective Measures to be Taken if Material is Released or Spilled</b>	Stop flow of material, contain/absorb small spills with dry sand, earth, and vermiculite. Do not use combustible materials. If possible, carefully neutralize spilled acid with soda ash, sodium bicarbonate, lime, etc. Wear acid-resistant clothing, boots, gloves, and face shield. Do not allow discharge of un-neutralized acid to sewer. Acid must be managed in accordance with approved local, state, and federal requirements. Consult state environmental agency and/or federal EPA.
<b>Waste Disposal Method</b>	Dispose of as a hazardous waste. Dispose of in accordance with applicable local, state and federal regulations.

## 7. HANDLING AND STORAGE

<b>Handling &amp; Storage</b>	Store frost-free under roof; prevent short circuits. Do not store in sealed, unventilated areas. Seek agreement with local water authorities in case of larger quantities. Avoid overheating and charging. Do not use organic solvents or anything other than manufacturers recommended cleaners on the batteries. If batteries have to be stored in storage rooms, it is imperative that the instructions for use are observed.
<b>Charging:</b>	There is a possible risk of electric shock from charging equipment and from strings of series connected batteries, whether or not being charged. Shut-off power to chargers whenever not in use and before detachment of any circuit connections. Batteries being charged may generate and release flammable hydrogen gas. Charging space should be ventilated. Prohibit smoking and avoid creation of flames and sparks nearby. Wear face and eye protection when near batteries being charged.
<b>Other</b>	Follow Manufacturers Recommendations regarding maximum recommended currents and operating temperature range. Do not overcharge beyond the recommended upper charging voltage limit. Applying pressure or deforming the battery may lead to disassembly followed by eye, skin and throat irritation.

## 8. EXPOSURE CONTROLS / PERSONAL PROTECTION

### Occupational exposure limits

#### US OSHA Specifically Regulated Substances (29 CFR 1910.1001 – 1050)

Ingredient	CAS Number	Type	Value
Lead	7439-92-1	TWA	0.05 mg/m <sup>3</sup>

#### US OSHA Table Z-1 Limits for Air Contaminants (29CFR 1910.1000)

Ingredient	CAS Number	Type	Value
Sulfuric Acid Absorbed in Glass Fiber	7664-93-9	PEL	1 mg/m <sup>3</sup>

#### US ACGIH Threshold Limit Values

Ingredient	CAS Number	Type	Value	Form
Lead	7439-92-1	TWA	0.05 mg/m <sup>3</sup>	
Sulfuric Acid	7664-93-9	TWA	0.2 mg/m <sup>3</sup>	Thoracic Fractions

#### US NIOSH: Pocket Guide to Chemical Hazards

Ingredient	CAS Number	Type	Value
Lead	7439-92-1	TWA	0.05 mg/m <sup>3</sup>
Sulfuric Acid	7664-93-9	TWA	1 mg/m <sup>3</sup>

#### International Exposure Limits (mg/m<sup>3</sup>)

*Chemical & Common Name	Quebec PEV	Ontario OEL	EU OEL
Lead and Lead Compounds (inorganic)	0.05	0.05	0.15 (a)
Electrolyte (H <sub>2</sub> SO <sub>4</sub> /H <sub>2</sub> O)	1	0.2	0.05 (b)

(a) As inhalable aerosol (b) Thoracic fraction

## Biological limit values

### ACGIH Biological Exposure Indices

Ingredient	Value	Determinant	Specimen	Sampling Time
Lead	300 µg/l	Lead	Blood	*

\* - For Sampling details please see the source document.

### Exposure Guidelines:

The OELs listed above are only applicable if the internal components of the battery cell are released. Follow standard monitoring procedures.

### Engineering Controls (Ventilation):

Store sealed lead acid batteries at ambient temperature. Never recharge batteries in an unventilated, enclosed space. Do not subject product to open flame or fire. Avoid conditions that could cause arcing between terminals.

### Respiratory Protection (NIOSH/MSHA approved):

NONE REQUIRED FOR NORMAL HANDLING OF THE FINISHED PRODUCT.

When concentrations of sulfuric acid mist are known to exceed PEL, use NIOSH or MSHA-approved respiratory protection.

### Skin Protection:

NONE REQUIRED FOR NORMAL HANDLING OF THE FINISHED PRODUCT.

If battery case is damaged, use rubber or plastic acid-resistant gloves with elbow-length gauntlet, acid-resistant apron, clothing and boots.

### Eye Protection:

NONE REQUIRED FOR NORMAL HANDLING OF THE FINISHED PRODUCT.

If necessary to handle damage product where exposure to the organic electrolyte is a possibility, chemical splash goggles and a face shield are recommended.

### Other Protection:

Safety footwear meeting the requirements of ANSI Z 41.1 is recommended when it is necessary to handle the finished product.

### General Hygiene Considerations:

When using, do not eat, drink, or smoke. Wash hands after handling. Contaminated work clothing should not be allowed out of the workplace. Handle in accordance with good industrial hygiene and safety practice.

## 9. PHYSICAL AND CHEMICAL PROPERTIES

<b>Appearance and Odor</b>	Manufactured article; no apparent odor.	
<b>Odor Threshold</b>	Not applicable.	
<b>pH</b>	Not applicable	
<b>Melting Point</b>	Lead - 621.32 °F (327.4 °C)	
<b>Boiling Point</b>	Not applicable unless individual components exposed. Battery Electrolyte (Acid) - 230 - 233.6 °F (110 - 112 °C) Lead - 3191 °F (1755 °C)	
<b>Flash Point</b>	Not applicable.	
<b>Evaporation Rate (Butyl Acetate = 1)</b>	Not applicable.	
<b>Vapor Pressure (mm Hg @ 20 ° C)</b>	Battery Electrolyte (Acid) 11.7	
<b>Flammability</b>		
<b>Upper/lower flammability or explosive limits</b>	Hydrogen	Flammability Limit Lower- 4.1 % Flammability Limit Upper – 74.2 %
<b>Vapor Pressure</b>	10.95 mm Hg (Sulfuric Acid)	
<b>Vapor Density</b>	Not applicable.	
<b>Relative Density</b>	1.21 - 1.3 Battery Electrolyte (Acid)	

<b>Solubility</b>	Lead and Lead dioxide are not soluble. 100 % Battery Electrolyte (Acid).
<b>% Volatile by Weight</b>	Not applicable unless individual components exposed.
<b>Partition coefficient (n-octanol/water)</b>	Not applicable
<b>Auto-ignition temperature</b>	Not applicable
<b>Decomposition temperature</b>	Not applicable
<b>Viscosity</b>	Not applicable
<b>Density</b>	11.35 g/cm <sup>3</sup> Lead

## 10. STABILITY AND REACTIVITY

<b>Reactivity</b>	This product is non-reactive under normal conditions or use, storage, and transport.
<b>Stability</b>	The sealed battery is considered stable.
<b>Conditions to Avoid</b>	Sparks and other sources of ignition; high temperature; over charging.
<b>Incompatibility (materials to avoid)</b>	Acid: Contact with combustibles and organic materials may cause fire and explosion. Also reacts violently with strong reducing agents, metals, sulfur trioxide gas, strong oxidizers, and water. Contact with metals may produce toxic sulfur dioxide fumes and may release flammable hydrogen gas.
<b>Hazardous Decomposition Products</b>	Lead compounds: Avoid contact with strong acids, bases, halides, halogenates, potassium nitrate, permanganate, peroxides, nascent hydrogen, and reducing agents. Acid: Sulfur trioxide, carbon monoxide, sulfuric acid mist, sulfur dioxide, hydrogen sulfide.
<b>Hazardous Polymerization</b>	Lead compounds: Temperatures above the melting point are likely to produce toxic metal fume, vapor, or dust; contact with strong acid or base or presence of nascent hydrogen may generate highly toxic arsine gas. Will not occur.

## 11. TOXICOLOGICAL INFORMATION

**NOTE: Under normal conditions of use, this product does not present a health hazard. The following information is provided for organic electrolyte and lead exposure that may occur due to container breakage or under extreme conditions such as fire. Organic electrolyte – reacts with moisture/water to produce hydrofluoric acid in trace quantities. Hydrofluoric acid is extremely corrosive and toxic. In severe exposures it acts as a systemic poison and causes severe burns. The reaction may be delayed. Any contact with this material, even minor, requires immediate medical attention.**

### ROUTES AND METHODS OF ENTRY

<b>Inhalation</b>	<b>EXPOSURE IS NOT EXPECTED FOR PRODUCT UNDER NORMAL CONDITIONS OF USE.</b>  Sulfuric Acid: Breathing of sulfuric acid vapors or mists may cause severe respiratory irritation. Lead Compounds: Inhalation of lead dust or fumes may cause irritation of upper respiratory tract and lungs.
<b>Skin Contact</b>	<b>EXPOSURE IS NOT EXPECTED FOR PRODUCT UNDER NORMAL CONDITIONS OF USE.</b>  Sulfuric Acid: Severe irritation, burns and ulceration. Lead Compounds: Not absorbed through the skin.
<b>Skin Absorption</b>	<b>EXPOSURE IS NOT EXPECTED FOR PRODUCT UNDER NORMAL CONDITIONS OF USE.</b>  In the event of overcharging or damage to the unit, exposure to organic electrolyte solution/mist is possible. Extreme exposures to the organic electrolyte can be absorbed through the skin.
<b>Eye Contact</b>	<b>EXPOSURE IS NOT EXPECTED FOR PRODUCT UNDER NORMAL CONDITIONS OF USE.</b> Sulfuric Acid: Severe irritation, burns, cornea damage, and blindness. Lead Compounds: May cause eye irritation.
<b>Ingestion</b>	<b>EXPOSURE IS NOT EXPECTED FOR PRODUCT UNDER NORMAL CONDITIONS OF USE.</b> Sulfuric Acid: May cause severe irritation of mouth, throat, esophagus and stomach.

Lead Compounds: Acute ingestion may cause abdominal pain, nausea, vomiting, diarrhea and severe cramping. This may lead rapidly to systemic toxicity and must be treated by a physician.

#### **SIGNS AND SYMPTOMS OF OVEREXPOSURE**

##### **Acute Effects**

**EXPOSURE IS NOT EXPECTED FOR PRODUCT UNDER NORMAL CONDITIONS OF USE.**

Sulfuric Acid: Severe skin irritation, damage to cornea, upper respiratory irritation.

Lead Compounds: Symptoms of toxicity include headache, fatigue, abdominal pain, loss of appetite, muscular aches and weakness, sleep disturbances and irritability

##### **Chronic Effects**

**EXPOSURE IS NOT EXPECTED FOR PRODUCT UNDER NORMAL CONDITIONS OF USE.**

Sulfuric Acid: Possible erosion of tooth enamel, inflammation of nose, throat & bronchial tubes.

Lead Compounds: Anemia; neuropathy, particularly of the motor nerves, with wrist drop; kidney damage; reproductive changes in males and females. Repeated exposure to lead and lead compounds in the workplace may result in nervous system toxicity. Some toxicologists report abnormal conduction velocities in persons with blood lead levels of 50 µg/100 ml or higher. Heavy lead exposure may result in central nervous system damage, encephalopathy and damage to the blood-forming (hematopoietic) tissues.

#### **MEDICAL CONDITIONS AGGRAVATED BY EXPOSURE**

Overexposure to sulfuric acid mist may cause lung damage and aggravate pulmonary conditions. Contact of sulfuric acid with skin may aggravate diseases such as eczema and contact dermatitis. Lead and its compounds can aggravate some forms of kidney, liver and neurologic diseases.

#### **ADDITIONAL HEALTH DATA**

All heavy metals, including the hazardous ingredients in this product, are taken into the body primarily by inhalation and ingestion. Most inhalation problems can be avoided by adequate precautions such as ventilation and respiratory protection covered in Section 8. Follow good personal hygiene to avoid inhalation and ingestion: wash hands, face, neck and arms thoroughly before eating, smoking or leaving the work site. Keep contaminated clothing out of non-contaminated areas, or wear cover clothing when in such areas. Restrict the use and presence of food, tobacco and cosmetics to non-contaminated areas. Work clothes and work equipment used in contaminated areas must remain in designated areas and never taken home or laundered with personal non-contaminated clothing. This product is intended for industrial use only and should be isolated from children and their environment.

The 19th Amendment to EC Directive 67/548/EEC classified lead compounds, but not lead in metal form, as possibly toxic to reproduction. Risk phrase 61: May cause harm to the unborn child, applies to lead compounds, especially soluble forms.

#### **Toxicological Data**

<b>Constituents</b>	<b>Species</b>	<b>Test Results</b>
Sulfuric Acid absorbed in glass-fiber material (CAS 7664-93-9)		
<b>Acute</b>		
<i>Oral</i>		
LD50	Rat	2140 mg/kg
<b>Skin corrosion/irritation</b>	Electrolyte: Causes severe skin burns	
<b>Serious eye damage/eye irritation</b>	Electrolyte: Causes severe skin burns	
<b>Respiratory Sensitization</b>	Not Classified	
<b>Skin Sensitization</b>	Not a skin sensitizer	
<b>Germ Cell Mutagenicity</b>	No data available	

#### **CARCINOGENICITY**

Under normal handling and storage conditions, the exposure to carcinogenic components is not expected. Risk of adverse effects occurs only if the cell is mechanically, thermally, or electrically abused to the point of compromising the enclosure.

Sulfuric Acid: The International Agency for Research on Cancer (IARC) has classified "strong inorganic acid mist containing sulfuric acid" as a Category I carcinogen, a substance that is carcinogenic to humans. This classification does not apply to liquid forms of sulfuric acid or sulfuric acid solutions contained within a battery. Inorganic acid mist (sulfuric acid mist) is not generated under normal use of this product. Misuse of the product, such as overcharging, may result in the generation of sulfuric acid mist.

Lead Compounds: Lead is listed as a 2B carcinogen, likely in animals at extreme doses. Proof of carcinogenicity in humans is lacking at present.

**IARC Monographs. Overall Evaluation of Carcinogenicity**

Lead (CAS 7439-92-1) 2B Possibly carcinogenic to humans.

**NTP Report on Carcinogens**

Lead sulfate (CAS 7446-14-2) Reasonably Anticipated to be a Human Carcinogen.

**OSHA Specifically Regulated Substances (29 CFR 1910.1001-1050)**

Not listed.

<b>Reproductive toxicity</b>	May damage fertility or the unborn child.
<b>Specific target organ toxicity - single exposure</b>	No data available.
<b>Specific target organ toxicity - repeated exposure</b>	Lead: May cause damage to organs (blood, central nervous system) through prolonged or repeated exposure.
<b>Aspiration hazard</b>	Not classified.

**12. ECOLOGICAL INFORMATION**

<b>Environmental Fate</b>	Lead is very persistent in soil and sediments. No data on environmental degradation. Mobility of metallic lead between ecological compartments is slow. Bioaccumulation of lead occurs in aquatic and terrestrial animals and plants but little bioaccumulation occurs through the food chain. Most studies include lead compounds and not elemental lead
<b>Ecotoxicity</b>	Very toxic to aquatic life with long lasting effects. However, no ecological impacts expected under normal use conditions.

Constituents	Species	Test Results
Inorganic Lead/Lead Compounds (CAS 7439-92-1)		
<b>Aquatic Fish</b>	LC50 Rainbow trout, Donaldson trout (Oncorhynchus mykiss)	1.17 mg/l, 96 hours
<b>Persistence and Degradability</b>	No data available	
<b>Bioaccumulative potential</b>	No data available	
<b>Additional Information</b>	No known effects on stratospheric ozone depletion Volatile organic compounds: 0% (by Volume) Water Endangering Class (WGK): NA	

**13. DISPOSAL CONSIDERATIONS**

<b>Waste disposal method</b>	Material should be recycled if possible. Lead-acid batteries are completely recyclable. Product can be recycled along with automotive (SLI) lead-acid batteries. Dispose waste and residues in accordance with applicable federal, state, and local regulations.
<b>Hazardous waste code</b>	D008: Lead
<b>Waste from residues / unused products</b>	Dispose of in accordance with local regulations. Empty containers or packaging may retain some product residues. This material and its container must be disposed of in a safe manner (see: Disposal instructions).
<b>Contaminated packaging</b>	Empty containers should be taken to an approved waste handling site for recycling or disposal.

**14. TRANSPORT INFORMATION**

**Note: Transportation requirements do not apply once the battery pack has been installed in a vehicle as part of the vehicle's functional components.**

**Transportation: Absorptive Glass-Fiber Material Lead Acid Battery is not a DOT Hazardous Material**





Lead ( CAS 7439-92-1)

**Clean Air Act (CAA) Section 112(r) Accidental Release Prevention (40 CFR 68.130)**

Sulfuric Acid absorbed in Glass-fiber Material (CAS 7664-93-9)

**Safe Drinking Water Act (SDWA)**

Not regulated

**Drug Enforcement Administration (DEA). List 2, Essential Chemicals (21 CFR 1310.02(b) and 1310.04(f)(2) and Chemical Code Number**

Sulfuric Acid absorbed in Glass-fiber Material (CAS 7664-93-9) 6552

**Drug Enforcement Administration (DEA). List 1 & 2 Exempt Chemical Mixtures (21 CFR 1310.12(c))**

Sulfuric Acid absorbed in Glass-fiber Material (CAS 7664-93-9) 20 % WV

**DEA Exempt Chemical Mixtures Code Number**

Sulfuric Acid absorbed in Glass-fiber Material (CAS 7664-93-9) 6552

**US State Regulations**

**US. Massachusetts RTK – Substance List**

Lead ( CAS 7439-92-1)  
Sulfuric Acid absorbed in Glass-fiber Material (CAS 7664-93-9)

**US New Jersey Worker and Community Right-to-know Act**

Lead ( CAS 7439-92-1)  
Sulfuric Acid absorbed in Glass-fiber Material (CAS 7664-93-9)

**US Pennsylvania Worker and Community Right-to-know Law**

Lead ( CAS 7439-92-1)  
Sulfuric Acid absorbed in Glass-fiber Material (CAS 7664-93-9)

**US Rhode Island RTK**

Lead ( CAS 7439-92-1)  
Sulfuric Acid absorbed in Glass-fiber Material (CAS 7664-93-9)

**US. California Proposition 65**

WARNING: This product contains chemicals known to the State of California to cause cancer. Battery posts, terminals and related accessories contain lead and lead compounds, chemicals known to the state of California to cause cancer and reproductive harm. Wash hands after handling.

\*Battery companies not party to the 1999 consent judgment with Mateel Environmental Justice Foundation should include a Proposition 65 Warning that complies with the current version of Proposition 65.

**US - California Proposition 65 - Carcinogens & Reproductive Toxicity (CRT): Listed substance**

Lead ( CAS 7439-92-1)  
Sulfuric Acid absorbed in Glass-fiber Material (CAS 7664-93-9)

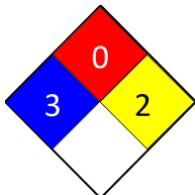
**International Inventories**

Country(s) or Region	Inventory Name	On inventory (yes/no)*
United States & Puerto Rico	Toxic Substances Control Act (TSCA) Inventory	Yes

\* A "Yes" indicates this product complies with the inventory requirements administered by the governing country(s). A "No" indicates that one or more components of the product are not listed or exempt from listing on the inventory administered by the governing country(s).

**16. OTHER INFORMATION**

**Issue Date:** 04/01/2015  
**Further information:** NFPA Hazard Scale: 0 = Minimal 1 = Slight 2 = Moderate 3=Serious 4 = Severe  
**NFPA ratings**



**Disclaimer**

Johnson Controls Battery Group, Inc. cannot anticipate all conditions under which this information and its product, or the products of other manufacturers in combination with its product, may be used. It is the user's responsibility to ensure safe conditions for handling, storage and disposal of the product, and to assume liability for loss, injury, damage or expense due to improper use. The information in the sheet was written based on the best knowledge and experience currently available.